



**JAI HIND COLLEGE  
BASANTSING INSTITUTE OF SCIENCE  
&  
J.T.LALVANI COLLEGE OF COMMERCE  
(AUTONOMOUS)**

"A" Road, Churchgate, Mumbai - 400 020, India.

**Affiliated to  
University of Mumbai**

Program: M.Sc. in Chemistry

Course: Analytical Chemistry II

Semester II

**Credit Based Semester and Grading System (CBSGS) with  
effect from the academic year 2021-22**



  
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## M.Sc. Chemistry Syllabus

Semester II			
Course Code	Course Title	Credits	Lectures/Week
PSCHE204	Analytical Chemistry II	04	04



  
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## Semester II – Theory

Course: PSCHE204	Course Title: Analytical Chemistry II	Credits: 04 Lectures/Week: 04
	<b>Chromatography, Thermal methods, automation in chemical analysis, Surface Analytical &amp; Electroanalytical methods</b>	
	<b>Objectives:</b> <ol style="list-style-type: none"> <li>1. To understand the concept and instrumentation of chromatography, thermal methods, automation in chemical analysis, surface analytical &amp; electroanalytical methods.</li> <li>2. To identify the application of Chromatography, Thermal methods, automation in chemical analysis, Surface Analytical &amp; Electroanalytical methods</li> </ol> <b>Outcomes:</b> <ol style="list-style-type: none"> <li>1. To explain the concept and instrumentation of chromatography, thermal methods, automation in chemical analysis, surface analytical &amp; electroanalytical methods.</li> <li>2. To illustrate the application of Chromatography, Thermal methods, automation in chemical analysis, Surface Analytical &amp; Electroanalytical methods.</li> </ol>	
Unit I	<b>Chromatography [15L]</b> <p><b>1.1 Basic concepts in chromatography:</b> Classification of chromatographic methods, theories of separation in chromatography; Optimization of chromatographic conditions. [3L]</p> <p><b>1.2 Gas Chromatography:</b> Instrumentation of GC with special reference to sample injection systems – split/splitless, column types, solid/ liquid stationary phases, column switching techniques, temperature programming, detectors in GC- Thermionic, thermal conductivity detector, Flame ionisation detector, mass spectrometric detector, Applications. [5 L]</p> <p><b>1.3 High Performance Liquid Chromatography (HPLC):</b> Normal phase and reversed phase with special reference to types of commercially available columns (Use of C8 and C18 columns). Detectors in HPLC: uv detector- Diode array type and fluorescence detector, mass detector, Applications of HPLC. Chiral and ion chromatography. [7 L]</p>	15L



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Unit II	<p><b>Thermal Methods, Hyphenation &amp; Automation in Chemical Analysis</b></p> <p><b>2.1 Thermal Methods [7L]</b></p> <p><b>2.1.1</b> Recapitulation of types of thermal methods, comparison between TGA, DTA and DSC.</p> <p><b>2.1.2</b> Applications of DSC - Heat of reaction, Specific heat, Safety screening, polymers, liquid crystals, Percentage crystallinity, oxidative stability, Drug analysis, Magnetic transition. e.g. Analysis of Polyethylene for its crystallinity.</p> <p><b>2.1.3</b> Enthalpimetric methods</p> <p><b>2.1.4</b> Thermometric titration</p> <p><b>2.1.5</b> Evolved Gas Analysis</p> <p><b>2.2 Hyphenated Techniques [3L]</b> Introduction, need for hyphenation, possible hyphenation, interfacing devices and applications of the following: GC-MS, GC-IR, MS-MS, LC-MS</p> <p><b>2.3 Automation in Chemical Analysis [4L]</b> Need for automation, objectives of automation, an overview of automated instruments and instrumentation, process control analysis, flow injection analysis, discrete automated systems, automatic analysis based on multi-layered films, gas monitoring equipment, Automatic titrators</p>	15L
Unit III	<p><b>Surface Analytical Techniques [15L]</b></p> <p><b>3.1 Introduction:</b> Principle, Types, Surface Sensitivity &amp; Specificity, Electron scattering, distinction of surface species [3L]</p> <p><b>3.2 Photoelectron Spectroscopy:</b> Principle, Instrumentation and Applications of X-ray photoelectron Spectroscopy (XPS, ESCA) &amp; Auger spectroscopy [3L]</p> <p><b>3.3 Electron Microscopy:</b> Principle, Instrumentation and Applications of Scanning Electron Microscopy (SEM) and Transmission Electron Microscopy (TEM) [4L]</p> <p><b>3.4 Scanning Probe Microscopy:</b> Principle, Instrumentation and Applications of Scanning Tunneling Microscopy (STM) and Atomic force Microscopy (AFM) [3L]</p> <p><b>3.5 Ion Spectroscopy:</b> Principle, Instrumentation and Applications of Ion Spectroscopy (ISS &amp; SIMS) [2L]</p>	15L



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<b>Unit IV</b>	<b>Electroanalytical Methods</b> <p><b>1.1 Ion Selective Potentiometry:</b> Ion selective electrodes and their applications (solid state, precipitate, liquid –liquid, enzyme and gas sensing electrodes), ion selective field effect transistors, biocatalytic membrane electrodes and enzyme based biosensors. [6L]</p> <p><b>1.2 Polarography:</b> Ilkovic equation, derivation starting with Cottrell equation, effect of complex formation on the polarographic waves. [4L]</p> <p><b>1.3 Electrogravimetry:</b> Introduction, principle, instrumentation, factors affecting the nature of the deposit, applications. [3L]</p> <p><b>1.4 Coulometry:</b> Introduction, principle, instrumentation, coulometry at controlled potential and controlled current. [2L]</p>	<b>15L</b>
<b>Standard References:</b> <p><b>Unit I</b></p> <ol style="list-style-type: none"> <li>1. Fundamentals of Analytical Chemistry, By Douglas A. Skoog, Donald M. West, F. James Holler, Stanley R. Crouch, 9<sup>th</sup> Edition, 2004.</li> <li>2. HPLC Practical and Industrial Applications, 2<sup>nd</sup> Ed., Joel K. Swadesh, CRC Press</li> </ol> <p><b>Unit II</b></p> <ol style="list-style-type: none"> <li>3. Instrumental Analysis, 5<sup>th</sup> Edition, Skoog, Holler and Nieman.</li> <li>4. Handbook of Thermal Analysis- Hatakeyama &amp; Liu, Wiley.</li> <li>5. Hyphenated techniques in Speciation Analysis- Royal Society of Chemistry (2003) Joanna Sepunar and Ryszard Lobinski.</li> </ol> <p><b>Unit III</b></p> <ol style="list-style-type: none"> <li>6. Modern techniques of surface science by D.P. Woodruff, T.A. Delchar, Cambridge Univ. Press, 1994.</li> </ol> <p><b>Unit IV</b></p> <ol style="list-style-type: none"> <li>7. Electrochemical Methods Fundamentals and Applications, Allen J Bard and Larry R Faulkner, John Wiley and Sons, (1980)</li> <li>8. Instrumental Methods of Analysis Willard, Merrit, Dean and Settle, 7<sup>th</sup> edition, CBS publishers.</li> <li>9. Principles of Instrumental Analysis – Skoog, Holler, Nieman, 5<sup>th</sup> Edition, Harcourt College Publishers, 1998. Chapters - 23, 24, 25.</li> </ol> <p><b>Additional References:</b></p> <ol style="list-style-type: none"> <li>10. Physical Principles of Electron Microscopy, An Introduction to TEM, SEM, and AEM Authors: Ray F. Egerton, ISBN: 978-0- 387-25800- 3 (Print)</li> <li>11. Introduction to Scanning Tunneling Microscopy by C. J. Chen, Oxford University Press, New York, 1993</li> </ol>		



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12. Transmission Electron Microscopy: A text book for Material Science, David B Williams and C., Barry Carter, Springer
13. Modern Spectroscopy, by J.M. Hollas, 3rd Edition (1996), John Wiley, New York
14. Analytical Chemistry Principles – John H Kennedy, 2nd edition, Saunders College Publishing (1990)
15. Modern Analytical Chemistry David Harvey; McGraw Hill Higher education publishers, (2000).

### Evaluation Scheme

- **Continuous Assessment (CA) – 40 Marks**
  - Knowledge and Application based: Objective test of 20 Marks
  - Skill based (20 marks): Learner will be assessed on relevant skills pertaining to the course content of a particular paper which could involve but not limited to
    - Oral Presentations on relevant topics
    - Review writing/Worksheets etc.
- **Semester End Examination (SEE)- 60 Marks**



  
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Program: M.Sc. in Chemistry

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Semester II

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## M.Sc. Chemistry Practical Syllabus

Semester II			
Course Code	Course Title	Credits	Practical/Week
PSCHEPR204	Analytical Chemistry Practical II	02	01




  
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## Semester II – Practical

<b>Course Code:</b> PSCHEPR204	<b>Course Title:</b> Analytical Chemistry Practical II	<b>Credits:</b> 02, <b>Practical/Week:</b> 01
	<b>Objectives:</b> <ol style="list-style-type: none"> <li>1. To develop the skill of determination of d-block elements from solutions mixtures.</li> <li>2. To acquainted with procedures of standardisation of various instruments.</li> </ol> <b>Outcomes:</b> <ol style="list-style-type: none"> <li>1. To determine d-block elements from solutions mixtures using different colorimetric instruments.</li> <li>2. To understand spectrometry and practical training of determining the analyte from sample matrix.</li> </ol>	
	<ol style="list-style-type: none"> <li>1. To determine percentage purity of sodium carbonate in washing soda pH metrically.</li> <li>2. To determine the amount of Ti(III) and Fe(II) in a mixture by titration with Ce(IV) potentiometrically.</li> <li>3. To determine the percentage purity of a sample (glycine/sodium benzoate/primary amine) by titration with perchloric acid in a non aqueous medium using glass calomel system potentiometrically.</li> <li>4. To determine the amount of nitrite present in the given water sample colorimetrically.</li> <li>5. To determine the amount of Fe(II) and Fe(III) in a mixture using 1,10-phenanthroline spectrophotometrically.</li> <li>6. Simultaneous determination of Cr(VI) and Mn(VII) in a mixture spectrophotometrically.</li> <li>7. To determine the percentage composition of HCl and H<sub>2</sub>SO<sub>4</sub> on weight basis in a mixture of two by conductometric titration with NaOH and BaCl<sub>2</sub>.</li> <li>8. To determine the amount of potassium in the given sample of fertilizers using a flame photometer by standard addition method.</li> </ol> <p style="text-align: center;">REFERENCES:</p> <ol style="list-style-type: none"> <li>1. Quantitative Inorganic Analysis including Elementary Instrumental Analysis by A. I. Vogels, 3<sup>rd</sup> Ed. ELBS (1964)</li> <li>2. Vogel's textbook of quantitative chemical analysis, Sixth Ed. Mendham, Denny, Barnes, Thomas, Pearson education</li> <li>3. Standard methods of chemical analysis, F. J. Welcher</li> <li>4. Standard Instrumental methods of Chemical Analysis, F. J. Welcher</li> <li>5. W.W.Scott."Standard methods of Chemical Analysis",Vol.I, Van Nostrand Company,Inc.,1939.</li> <li>6. E.B.Sandell and H.Onishi,"Spectrophotometric Determination of Traces of Metals",Part- II,4th Ed.,A Wiley Interscience Publication,New York,1978.</li> </ol>	




  
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## Evaluation Scheme

- Semester End Examination (SEE)- 50 Marks



  
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